

II. The Wavefunction and the Wave Equation

▪ Background:

- Need wave description for particle(s)
- To proceed, introduced wavefunction $\Psi(x,t)$ or $\Psi(\vec{r},t)$
- $|\Psi(x,t)|^2 dx = \text{Prob. of finding particle in interval } x \text{ to } x+dx \text{ at time } t$

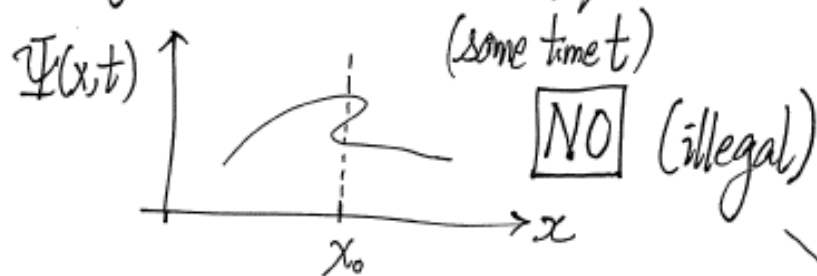
Questions[†]:

- What do we expect $\Psi(x,t)$ to behave based on physics?
- If we somehow know $\Psi(x,t)$, how to use it for calculations?
- What is the governing wave equation for $\Psi(x,t)$?

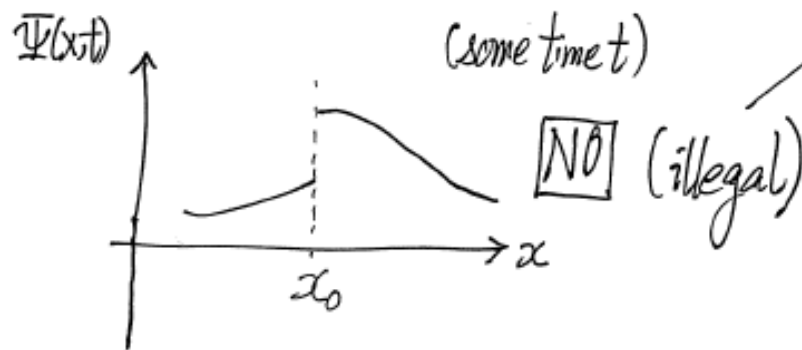
[†] We use $\Psi(x,t)$ here for easy visualization. Discussions also are valid for $\Psi(\vec{r},t)$.

A. Some expected behavior of $\Psi(x,t)$ based on physics

▪ Single-value :



▪ Continuous :



Can't tell probability[†] of finding particle at x_0

▪ Can Ψ be negative? YES!

▪ Can Ψ be complex? YES!

Physical Meaning is attached to $|\Psi(x,t)|^2$

[†] Here, we use physical sense. Continuity requirement can also be shown mathematically, after knowing the wave equation. Single-valuedness applies almost in all cases (except strange spin situations).

B. Normalization

▪ Formal statement

- $|\Psi(x,t)|^2 dx = \text{Prob. of finding particle in } x \rightarrow x+dx \text{ at time } t$
- Particle is there and cannot be destroyed (or more being created)

Ex. H-atom



"know" electron
to be around

$\Psi(\vec{r}, t)$ describes
state of electron


$$\int_{-\infty}^{\infty} |\Psi(x,t)|^2 dx = 1 \quad \text{or} \quad \int_{\text{all space}} |\Psi(\vec{r}, t)|^2 d^3r = 1 \quad [\text{Eq. (1)}]$$

Practically, it means that $\int_{-\infty}^{\infty} |\Psi(x,t)|^2 dx = \text{finite}$, then $\Psi(x,t)$ can be normalized

Practical Tips

- In a large class of QM problems on studying bound states [states with $\psi \rightarrow 0$ as $x \rightarrow \pm\infty$ (fast enough)], the wavefunctions of bound states can be normalized.


E.g.: all states in an infinite well & harmonic oscillator

states with energies lower than $V(\pm\infty)$ as in 

and in atomic states with energy < 0


- ∴ We will make use of normalization requirement in bound states in
 - 1D infinite and finite wells, 1D harmonic oscillator
 - 2D wells, rotor
 - 3D wells, Hydrogen atom, other atoms, molecules

- But there are solutions of Schrödinger Equation that are NOT bound states!

E.g.  a state with $E > V$ (called scattering state)

- These states cannot be normalized as in Eq. (1)
- But these states are sometimes useful

What to do then? There are several ways out!

- Use wave packet  to describe particle
 formed by adding up unbound states (c.f. Fourier) localized in a range of space

- "Normalized" them differently (not as Eq. (1))
- Let space be big but finite (often use in Solid State Physics)

▪ Key Points

- In handling bound states in QM, need to consider normalization
- Don't worry too much about unbound states, until it is absolutely necessary.

▪ Appreciation

- In classical physics, $\vec{E}(x,t)$ and $10 \cdot \vec{E}(x,t)$ are different,
different intensities
- In QM bound states, $\Psi(x,t)$ and $10 \cdot \Psi(x,t)$ are the "same".

They give the same information on prob. at x to $x+dx$. Only one is normalized and another is not properly normalized (but can be normalized)

$$\int_{-\infty}^{\infty} |\bar{\Psi}^{\text{not normalized}}|^2 dx = A \text{ (finite)} \quad [\text{still fine, no problem}]$$

Construct: $\Psi(x,t) = \frac{1}{\sqrt{A}} \bar{\Psi}^{\text{not normalized}}$

[This step is called normalization]

this is now properly normalized as

$$\int_{-\infty}^{\infty} |\Psi(x,t)|^2 dx = \frac{1}{A} \int_{-\infty}^{\infty} |\bar{\Psi}^{\text{not normalized}}|^2 dx = \frac{1}{A} \cdot A = 1$$

As long as A is **finite** (not infinite), the wavefunction can be normalized.

Summary: For a normalized wavefunction

$$|\Psi(x,t)|^2 dx = \text{Prob. of finding particle in } x \rightarrow x+dx \text{ at time } t$$

and a similar interpretation in 2D and 3D

▪ Probabilistic interpretation:

Max Born [Nobel Prize 1954]

Heisenberg was Born's RA after graduation. During which Heisenberg, Born, Kramer, Jordan wrote the first papers on QM in 1924-25. Heisenberg was awarded the 1932 Nobel Prize (Born was left out).



Max Born (1882-1970)

Other contributions include:

- Mentored Heisenberg even before he graduated from Munich with a thesis on turbulence (with Sommerfeld)
- Took Heisenberg's idea to Hilbert (Born and Hilbert were both at Gottingen), helped establish the mathematical foundation of QM
- Born-Oppenheimer approximation (molecular physics)
- Born-Huber cycle (thermodynamics, chemistry)
- Born approximation (QM scattering theory)
- Born and (Kun) Huang (lattice vibrations in solid state physics) [Kun Huang trained generations of semiconductor physicists in China]